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## Structure Reports

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## Key indicators

Single-crystal X-ray study
$T=292 \mathrm{~K}$
Mean $\sigma(\mathrm{C}-\mathrm{C})=0.003 \AA$
Disorder in main residue
$R$ factor $=0.056$
$w R$ factor $=0.152$
Data-to-parameter ratio $=14.8$

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.
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## Propyl 3,5-dinitrobenzoate

The title compound, $\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{6}$, has two independent molecules in the asymmetric unit. All bond lengths and angles in the molecules are in normal ranges. The crystal packing is stabilized by van der Waals forces and $\pi-\pi$ stacking interactions.

## Comment

Due to their biological activities, dinitrobenzoate derivatives are widely used in pharmacology. It is known (Olive, 1979; Huang et al., 2004) that synthetic dinitrobenzoate compounds are useful in the synthesis of DNA and oligosaccharides. Dinitrobenzoic acid derivatives are effective in tumour treatment as radiation sensitizers (Kagitani et al., 1984). Furthermore, a recently synthesized series of 3,5-dinitrobenzoic acid esters has demonstrated promising anticreatinine effects (Yu \& Yang, 2002). As part of an ongoing study of dinitrobenzoate derivatives, we report here the crystal structure of the title compound, (I).

(I)

All bond lengths and angles in the two independent molecules of (I) (Fig. 1) are normal (Table 1). Except for methyl atoms C10 and C20, all non-H atoms of each molecule lie in a plane (r.m.s deviations $=0.037$ and $0.094 \AA$ A respectively). The


Figure 1
The two independent molecules of (I) with the disordered methyl group in one (both components shown). Displacement ellipsoids are drawn at the $30 \%$ probability level.

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crystal packing is stabilized by van der Waals forces and $\pi-\pi$ stacking interactions, evidenced by the short intermolecular contacts N3‥O8 ${ }^{\mathrm{i}}$ [3.045 (3) A ; symmetry code (i): $2-x$, $2-y, \quad 1-z], \quad \mathrm{C} 12 \cdots \mathrm{C} 12^{\mathrm{i}} \quad[3.436(3) \AA]$ and $\mathrm{C} 17 \cdots \mathrm{O} 9^{\mathrm{i}}$ [3.104 (3) Å].

## Experimental

The title compound, (I), was synthesized according to the literature procedure of Bartlett \& Trachtenberg (1958). Crystals suitable for X-ray analysis were grown from a solution in methanol at room temperature by slow evaporation.

## Crystal data

$\mathrm{C}_{10} \mathrm{H}_{10} \mathrm{~N}_{2} \mathrm{O}_{6}$
$M_{r}=254.20$
Triclinic, $P \overline{1}$
$a=11.0566(16) \AA$
$b=11.3596(17) \AA$
$c=11.4916(17) \AA$
$\alpha=117.554(2)^{\circ}$
$\beta=99.937(3)^{\circ}$
$\gamma=102.830(2)^{\circ}$
$V=1182.7(3) \AA^{\circ}$

$$
\begin{aligned}
& Z=4 \\
& D_{x}=1.428 \mathrm{Mg} \mathrm{~m}^{-3}
\end{aligned}
$$

Mo $K \alpha$ radiation
Cell parameters from 2196
reflections
$\theta=2.2-25.6^{\circ}$
$\mu=0.12 \mathrm{~mm}^{-1}$
$T=292$ (2) K
Block, colourless
$0.60 \times 0.50 \times 0.30 \mathrm{~mm}$

## Data collection

Bruker SMART CCD area-detector diffractometer
$\varphi$ and $\omega$ scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
$T_{\text {min }}=0.931, T_{\text {max }}=0.965$
7233 measured reflections

## Refinement

Refinement on $F^{2}$
$R\left[F^{2}>2 \sigma\left(F^{2}\right)\right]=0.056$
$w R\left(F^{2}\right)=0.152$
$S=0.97$
4995 reflections
338 parameters


Figure 2
Packing diagram for (I). Only the major component of the disordered methyl group is shown.

H atoms were positioned geometrically and refined as riding, with $\mathrm{C}-\mathrm{H}=0.93-0.97 \AA$ and $U_{\text {iso }}(\mathrm{H})=1.2-1.5 U_{\text {eq }}(\mathrm{C})$. The methyl group attached to C20 was treated as disordered over two positions; the refined occupancies were 0.83 (2) and 0.17 (2).

Data collection: SMART (Bruker, 2001); cell refinement: SAINTPlus (Bruker, 2001); data reduction: SAINT-Plus; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: SHELXTL-NT (Bruker, 2001); software used to prepare material for publication: SHELXTL-NT.

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## References

Bartlett, P. D. \& Trachtenberg, E. N. (1958). J. Am. Chem. Soc. 80, 5808-5812. Bruker (2001). SMART (Version 5.628), SAINT-Plus (Version 6.45) and SHELXTL-NT (Version 6.12). Bruker AXS Inc., Madison, Wisconsin, USA.
Huang, H. Q., Li, Q., Cai, M. Sh. \& Li, Zh. J. (2004). J. Chin. Pharm. Sci. 13, 242-244.
Kagitani, T., Minagawa, M., Nakahara, Y,. Kimura, S., Tsubakimoto, T., Oshiumi, R. \& Sakano, K. (1984). Jpn Kokai Tokyo Koho JP 60258116.
Olive, P. L. (1979). Br. J. Cancer, 40, 89-93.
Sheldrick, G. M. (1996). SADABS. University of Göttingen, Germany.
Sheldrick, G. M. (1997). SHELXS97 and SHELXL97. University of Göttingen, Germany.
Yu, J. G. \& Yang, D. Z. (2002). Chin. J. Appl. Chem. 19, 858-861.

